

=> d 14 1-11 abs,bib

L4 ANSWER 1 OF 11 HCPLUS COPYRIGHT 2008 ACS on STN  
AB Novel three-dimensional (3D) hierarchical nanoarchitectures of  $\epsilon$ -MnO<sub>2</sub> have been synthesized by a simple chemical route without the addition of any surfactants or organic templates. The self-organized crystals consist of a major  $\epsilon$ -MnO<sub>2</sub> dipyramidal single crystal axis and six secondary branches, which are arrays of single-crystal  $\epsilon$ -MnO<sub>2</sub> nanobelts. The growth directions of the nanobelts are perpendicular to the central dipyramidal axis, which shows sixfold symmetry. The shape of the  $\epsilon$ -MnO<sub>2</sub> assembly can be controlled by the reaction temperature. The morphol. of  $\epsilon$ -MnO<sub>2</sub> changes from a six-branched star-like shape to a hexagonal dipyramidal morphol. when the temperature is increased from 160 to 180°C. A possible growth mechanism is proposed. The synthesized  $\epsilon$ -MnO<sub>2</sub> shows both semiconducting and magnetic properties. These materials exhibit ferromagnetic behavior below 25 K and paramagnetic behavior above 25 K. The  $\epsilon$ -MnO<sub>2</sub> system may have potential applications in areas such as fabrication of nanoscale spintronic materials, catalysis, and sensors.

AN 2006:270788 HCPLUS

DN 145:251620

TI Hydrothermal growth of manganese dioxide into three-dimensional hierarchical nanoarchitectures

AU Ding, Yun-Shuang; Shen, Xiong-Fei; Gomez, Sinue; Luo, Hong; Aindow, Mark; Suib, Steven L.

CS Institute of Materials Science, University of Connecticut, Storrs, CT, 06269, USA

SO Advanced Functional Materials (2006), 16(4), 549-555

CODEN: AFMDC6; ISSN: 1616-301X

PB Wiley-VCH Verlag GmbH & Co. KGaA

DT Journal

LA English

RE.CNT 51 THERE ARE 51 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 2 OF 11 HCPLUS COPYRIGHT 2008 ACS on STN

AB Quartz crystals synthesized hydrothermally above the high-low transition temperature showed various hexagonal dipyramidal morphologies mostly with and rarely without prism faces, which are sometimes curved leading to a spindle-shape. The development of prism faces is distinctly different from simple hexagonal dipyramidal morphol. without prism faces commonly exhibited by natural high quartz crystals. Thus the hitherto believed representative morphol. of natural high quartz is a habit appearing when high quartz crystals grew in silicate systems.

AN 1995:751623 HCPLUS

DN 123:156799

OREF 123:27675a,27678a

TI Growth and morphology of quartz crystals synthesized above the transition temperature

AU Hosaka, Masahiro; Miyata, Takeshi; Sunagawa, Ichiro

CS Yamanashi Institute of Gemmology and Jewelry Arts, Tokoji-machi 1955-1, Kofu, 400, Japan

SO Journal of Crystal Growth (1995), 152(4), 300-6

CODEN: JCRGAE; ISSN: 0022-0248

PB North-Holland

DT Journal

LA English

L4 ANSWER 3 OF 11 HCPLUS COPYRIGHT 2008 ACS on STN  
AB Prismatic crystals of jeremejevite (jv) [12322-87-1] are described from granitic pegmatites, in southwestern Pamirs, developed in contact with garnet-biotite banded gneisses. Crystals of jv, with a hexagonal and hexagonal-dipyramidal habit, are associated with apatite, schorl, biotite, lepidolite, beryl, etc. The d. 3.26 g/cm<sup>3</sup>, hardness 1251 N/mm<sup>2</sup>, ns  $\gamma$  = 1.646 and  $\alpha$  = 1.637, birefringence 0.009, and 2V = <44° of jv were determined. The unit-cell parameters are a 8.544 and c 8.159 Å; c/a = 1:1.047, and cell volume 515.80 Å<sup>3</sup>. Data show that jv crystallized from low-temperature hydrothermal solns. of high alkalinity, during the terminal stages of pegmatite-forming processes. The jv formation was related to excess amts. of K, Al, and B in residual pegmatitic fluids.

AN 1983:425566 HCPLUS

DN 99:25566

OREF 99:4081a, 4084a

TI Jeremejevite as a mineral recently found in the USSR

AU Konovalenko, S. I.; Rossovskii, L. N.; Anan'ev, S. A.

CS Krasnoyarsk. Otd., Sib. Nauchno-Issled. Inst. Geol., Geofiz. Mineral.

Syr'ya, Krasnoyarsk, USSR

SO Zapiski Vsesoyuznogo Mineralogicheskogo Obshchestva (1983), 112(2), 212-17  
CODEN: ZVMOAG; ISSN: 0044-1805

DT Journal

LA Russian

L4 ANSWER 4 OF 11 HCPLUS COPYRIGHT 2008 ACS on STN

AB Hexagonal, dipyramidal-prismatic, and isometric crystals of apatite [64476-38-6] from the title rocks have ns  $\omega$  = 1.640-1.651 and  $\epsilon$  = 1.635-1.646; chemical, these apatites are fluorapatites. Quant. estns. were made of La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tu, Yb, Lu, and Y. The heavy lanthanides are conspicuously absent.

AN 1979:560516 HCPLUS

DN 91:160516

OREF 91:25871a, 25874a

TI Rare earth elements and yttrium in accessory apatite of basic and ultrabasic rocks of the Voronezh crystalline massif

AU Plaksenko, A. N.

CS Voronezh. Gos. Univ., Voronezh, USSR

SO Geokhimiya (1979), (8), 1254-8

CODEN: GEOKAQ; ISSN: 0016-7525

DT Journal

LA Russian

L4 ANSWER 5 OF 11 HCPLUS COPYRIGHT 2008 ACS on STN

AB The hardness of hematite from the Korushnovsk, Shabarovsk, Olenogorsk, and the Krivoi-Rog ore deposits were determined on the pinacoidal (0001), rhombohedral (10.hivin.11), and hexagonal dipyramidal (22.hivin.43) faces; the average values for these 3 faces were, 1176, 1269, and 1290 kg/mm<sup>2</sup>, resp. Wide variations in the hardness of hematite are attributed to differences in crystal morphol.

AN 1975:430791 HCPLUS

DN 83:30791

OREF 83:4917a, 4920a

TI Anisotropy of the hardness of hematite

AU Petrova, L. V.; Domnina, L. I.

CS "Mekhanobr", Leningrad, USSR

SO Zapiski Vsesoyuznogo Mineralogicheskogo Obshchestva (1975), 104(1), 84-5  
CODEN: ZVMOAG; ISSN: 0044-1805

DT Journal

LA Russian

L4 ANSWER 6 OF 11 INSPEC (C) 2008 IET on STN  
AN 2006:8933879 INSPEC  
AB Novel three-dimensional (3D) hierarchical nanoarchitectures of X<sub>220</sub>A-MnO<sub>2</sub> have been synthesized by a simple chemical route without the addition of any surfactants or organic templates. The self-organized crystals consist of a major X<sub>220</sub>A-MnO<sub>2</sub> dipyramidal single crystal axis and six secondary branches, which are arrays of single-crystal X<sub>220</sub>A-MnO<sub>2</sub> nanobelts. The growth directions of the nanobelts are perpendicular to the central dipyramidal axis, which shows sixfold symmetry. The shape of the X<sub>220</sub>A-MnO<sub>2</sub> assembly can be controlled by the reaction temperature. The morphology of X<sub>220</sub>A-MnO<sub>2</sub> changes from a six-branched star-like shape to a hexagonal dipyramidal morphology when the temperature is increased from 160 to 180°C. A possible growth mechanism is proposed. The synthesized X<sub>220</sub>A-MnO<sub>2</sub> shows both semiconducting and magnetic properties. These materials exhibit ferromagnetic behavior below 25 K and paramagnetic behavior above 25 K. The X<sub>220</sub>A-MnO<sub>2</sub> system may have potential applications in areas such as fabrication of nanoscale spintronic materials, catalysis, and sensors  
AN 2006:8933879 INSPEC  
TI Hydrothermal growth of manganese dioxide into three-dimensional hierarchical nanoarchitectures  
AU Yun-Shuang Ding; Xiong-Fei Shen; Gomez, S.; Hong Luo; Aindow, M.; Suib, S.L. (Inst. of Mater. Sci., Connecticut Univ., Storrs, CT, USA)  
SO Advanced Functional Materials (3 March 2006), vol.16, no.4, p. 549-55, 31 refs.  
CODEN: AFMDC6, ISSN: 1616-301X  
SICI: 1616-301X(20060303)16:4L.549:HGMD;1-X  
Published by: Wiley-VCH, Germany  
DT Journal  
TC Experimental  
CY Germany  
LA English

L4 ANSWER 7 OF 11 INSPEC (C) 2008 IET on STN  
AN 1995:5054090 INSPEC DN A1995-20-8110D-010  
AB Quartz crystals synthesized hydrothermally above the high-low transition temperature showed various hexagonal dipyramidal morphologies mostly with and rarely without prism faces, which are sometimes curved leading to a spindle-shape. The development of prism faces is distinctly different from simple hexagonal dipyramidal morphology without prism faces commonly exhibited by natural high quartz crystals. Thus the hitherto believed representative morphology of natural high quartz is a habit appearing when high quartz crystals grew in silicate systems  
AN 1995:5054090 INSPEC DN A1995-20-8110D-010  
TI Growth and morphology of quartz crystals synthesized above the transition temperature  
AU Hosaka, M.; Miyata, T.; Sunagawa, I. (Inst. of Gemmology & Jewelry Arts, Yamanashi Univ., Kofu, Japan)  
SO Journal of Crystal Growth (July 1995), vol.152, no.4, p. 300-6, 9 refs.  
CODEN: JCRCGA, ISSN: 0022-0248  
Price: 0022-0248/95/\$09.50  
DT Journal  
TC Experimental  
CY Netherlands  
LA English

L4 ANSWER 8 OF 11 INSPEC (C) 2008 IET on STN  
AN 1989:3447361 INSPEC DN A1989-111197

AB Section Z of Meishan in Changxing of Zhejiang, known as the stratotype section of the Permian-Triassic boundary in South China, is also one of the candidate stratotype sections of the same boundary in the world. Here the strata crossing the Permian-Triassic boundary are well exposed. The Upper Permian Changhsingian is composed of limestones intercalated with thin-bedded claystone, while the lowermost part of the Lower Triassic is represented by mudstone, marl and clay. Recently, the writers first discovered a great number of quartz grains with hexagonal dipyratidal crystal form which is characteristic of high  $\beta$ -quartz in mixed Beds 1-3 of the uppermost part of Upper Permian and the lower-most part of Lower Triassic. High  $\beta$ -quartz always occurs in acidic volcanic rocks. Accordingly, the present discovery has not only revealed the influence of volcanic eruptions in the Changxing area during the late Late Permian and the early Early Triassic, but also provided new valuable data for the study of the event stratigraphy of the Permian-Triassic boundary

AN 1989:3447361 INSPEC DN A1989-111197

TI Discovery of paramorph of high-quartz in the stratotype section of the Permian-Triassic boundary at Meishan of Changxing, Zhejiang, and its significance

AU He Jin-Wen; (Nanjing Inst. of Geol. & Paleontology, Acad. Sinica, China), Chai Zhi-Fang; Ma Shu-Lan

SO Chinese Science Bulletin (March 1989), vol.34, no.6, p. 474-7, 2 refs.ISSN: 0250-7862

DT Journal

TC Experimental

CY China

LA English

L4 ANSWER 9 OF 11 INSPEC (C) 2008 IET on STN

AN 1983:1962525 INSPEC DN A1983-003735

AB By spontaneous crystallization from solution in a fluoride melt with slow fall of temperature the authors have obtained single crystals of oxygermanate-apatites (OGA) with the compositions Sr<sub>2</sub>Ln<sub>8</sub>Ge<sub>6</sub>O<sub>26</sub>, where Ln=La, Nd, Sm, Eu, Gd, or Dy. Chemical analyses of samples revealed that there is no deviation from stoichiometry during growth. The resulting single crystals are transparent, have a hexagonal-dipyramidal habit, and measure up to 20 mm along the hexagonal axis and up to 3 mm in cross section

AN 1983:1962525 INSPEC DN A1983-003735

TI Some properties of single crystals of oxygermanate-apatites

AU Fedorov, N.F.; Andreev, I.F.; Lukashov, I.L. (Technol. Inst., Leningrad, USSR)

SO Soviet Physics - Crystallography (March 1982), vol.27, no.2, p. 231, 5 refs.

CODEN: SPHCA6, ISSN: 0038-5638

Translation of: Kristallografiya (March 1982), vol.27, no.2, p. 384

CODEN: KRISAJ, ISSN: 0023-4761

DT Journal; Translation Abstracted

TC Experimental

CY United States; USSR

LA English

L4 ANSWER 10 OF 11 USPATFULL on STN

AB A corundum crystal formed body having a corundum crystal grown directly on a base material and a production process capable of producing the corundum crystal formed body easily at low costs. The a corundum crystal formed body has a platinum base material and a corundum crystal portion formed on the platinum base material. Further, the process for producing a corundum crystal formed body, involves forming a corundum

crystal is formed on a platinum base material by a flux evaporation method of heating a sample containing a raw material and a flux to precipitate a crystal. The crystal is grown by use of flux evaporation as a driving force.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AN 2007:161383 USPATFULL  
TI Corundum crystal formed body  
IN Teshima, Katsuya, Tokyo, JAPAN  
PI US 20070140936 A1 20070621  
AI US 2005-589546 A1 20050217 (10)  
WO 2005-JP2496 20050217  
20060816 PCT 371 date  
PRAI JP 2004-41834 20040218  
DT Utility  
FS APPLICATION  
LREP LADAS & PARRY LLP, 224 SOUTH MICHIGAN AVENUE, SUITE 1600, CHICAGO, IL, 60604, US  
CLMN Number of Claims: 21  
ECL Exemplary Claim: 1-15  
DRWN 4 Drawing Page(s)  
LN.CNT 1088

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

L4 ANSWER 11 OF 11 USPATFULL on STN

AB An artificial corundum crystal which can be put into practical use at low costs, and a process for producing the same. The artificial corundum crystal has at least one crystal face selected from {113}, {012}, {014}, {113}, {110}, {101}, {116}, {211}, {122}, {214}, {100}, {125}, {131}, and {312} faces. The process for producing the artificial corundum crystal is by a flux evaporation method of heating a sample containing a raw material and a flux to precipitate a crystal and grow the crystal by use of flux evaporation as a driving force.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

AN 2007:113551 USPATFULL  
TI Artificial corundum crystal  
IN Teshima, Katsuya, Tokyo-to, JAPAN  
Oishi, Shuji, Nagano, JAPAN  
PI US 20070098618 A1 20070503  
AI US 2004-581182 A1 20041130 (10)  
WO 2004-JP17753 20041130  
20060601 PCT 371 date  
PRAI JP 2003-402115 20030112  
JP 2004-20059 20040205  
JP 2004-144118 20040503  
JP 2004-144124 20040513  
JP 2004-144125 20040513  
DT Utility  
FS APPLICATION  
LREP LADAS & PARRY LLP, 224 SOUTH MICHIGAN AVENUE, SUITE 1600, CHICAGO, IL, 60604, US  
CLMN Number of Claims: 28  
ECL Exemplary Claim: 1  
DRWN 4 Drawing Page(s)  
LN.CNT 1452

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

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(FILE 'HOME' ENTERED AT 14:29:15 ON 05 SEP 2008)

FILE 'HCAPLUS, INSPEC, JAPIO, USPATFULL, USPATOLD, USPAT2' ENTERED AT  
14:40:51 ON 05 SEP 2008

L1        3536151 S (CRYSTAL#)  
L2        443126 S (AL2O3 OR CORUMDUM)  
L3        51960 S L1 AND L2  
L4        11 S L1 AND (HEXAGON?(3W)DIPYRAMIDAL?)  
L5        250923 S (COLORLESS OR NO(6A)COLOR?)  
L6        2 S L4 AND L5  
L7        961124 S (FLUX?)  
L8        1337 S L1 AND L2 AND L7  
L9        2 S L4 AND L7  
L10      20363 S (EVAPORAT?)(8A)(INHIBIT? OR ELIMINAT? OR RID? OR IRRADICAT? O  
L11      2 S L8 AND L10  
L12      2 S L1 AND L2 AND L7 AND L10

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